



Innovation in Green Hydrogen (and related technologies)

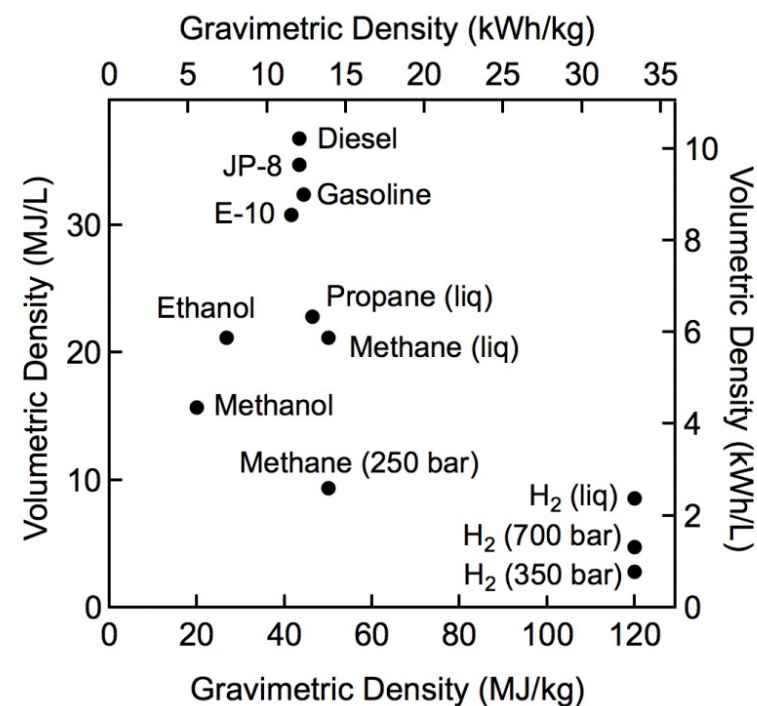
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Oxford Energy Day

28 September 2023



UK Hydrogen Strategy



This strategy sets out the approach to developing a thriving low carbon hydrogen sector in the UK to meet our increased ambition for 10GW of low carbon hydrogen production capacity by 2030. The page includes hydrogen strategy updates to the market.

<https://www.gov.uk/government/publications/uk-hydrogen-strategy>

Notice

Net Zero Hydrogen Fund strands 1 and 2: summaries of successful applicants round 1 (April 2022) competition

Published 30 March 2023

Trecwn Green Energy Hub

Ballymena Hydrogen

Conrad Energy Hydrogen
Lowestoft

Didcot Green Hydrogen
Electrolyser

Green Hydrogen St Helens

Green Hydrogen Winnington &
Middlewich

Inverness Green Hydrogen Hub

Mannok Green Hydrogen Valley

MCRU Integrated Hydrogen
Delivery for a Fuel Cell Van
Fleet Pilot

Lanarkshire Green Hydrogen

The Knockshinnoch Green
Hydrogen Hub Project

HyNet Hydrogen Production
Plant HPP2 **4 GW by 2030**

Kintore Hydrogen

H2NorthEast

Port of Felixstowe Green
Hydrogen Project

<https://www.gov.uk/government/publications/net-zero-hydrogen-fund-strands-1-and-2-successful-applicants>

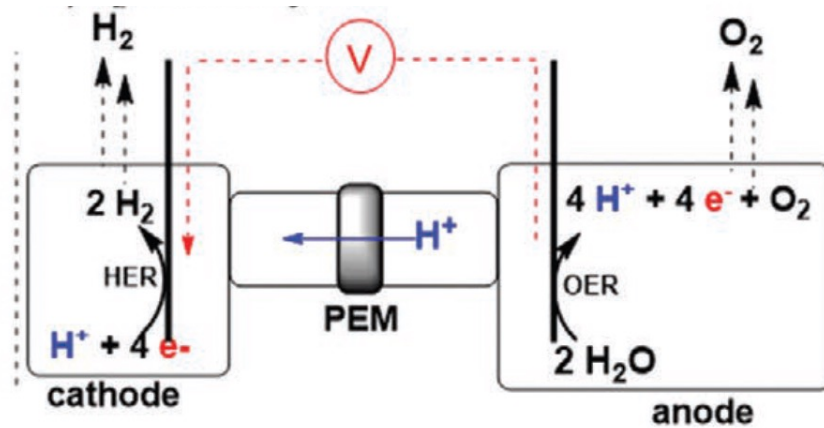
Green Hydrogen via Electrolysis

with Electricity from Renewables

Catalyst: Pt/C

Requirements:

Low overpotential
High current density



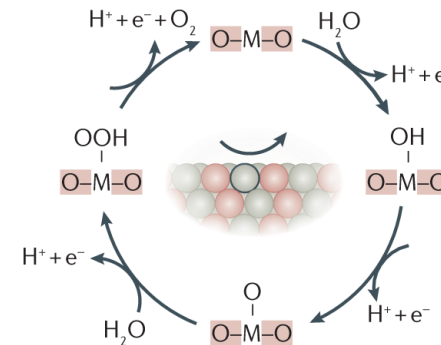
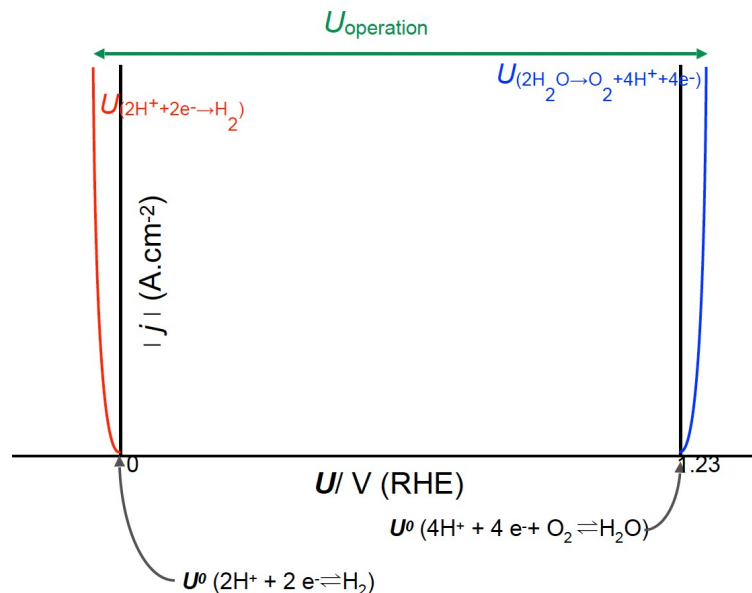
Catalyst: IrO₂

Requirements:

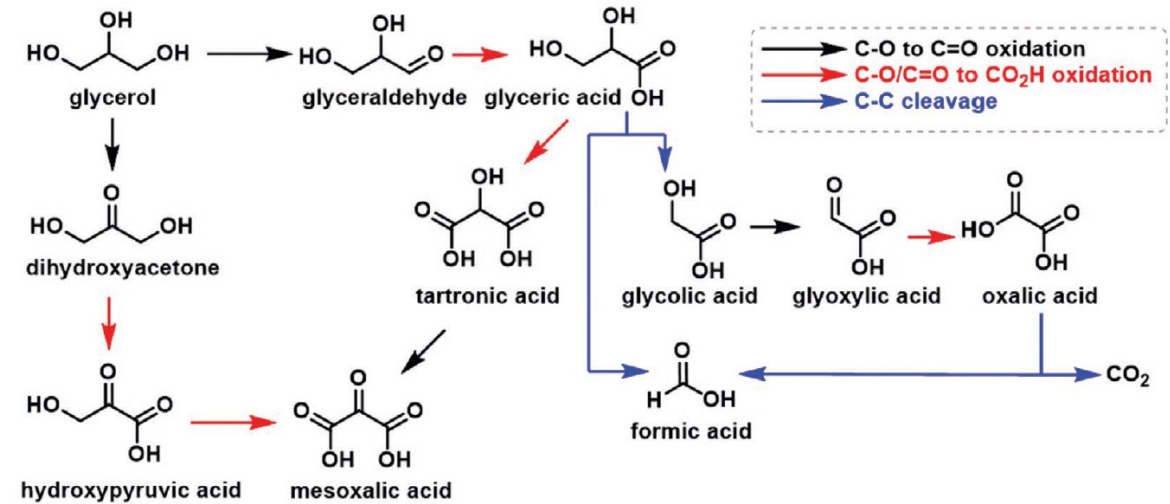
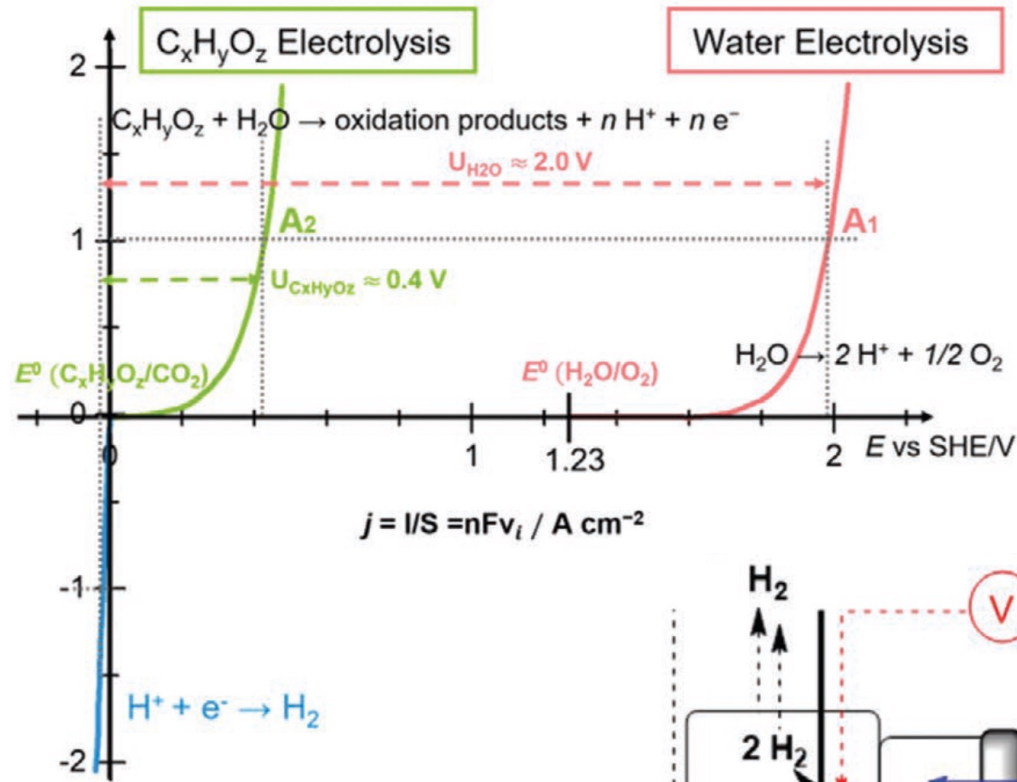
Low overpotential
High stability

Challenge 1: High overpotentials for OER (240 – 400 mV) → 1.7 – 2 V

Challenge 2: IrO₂ is too precious and scarce

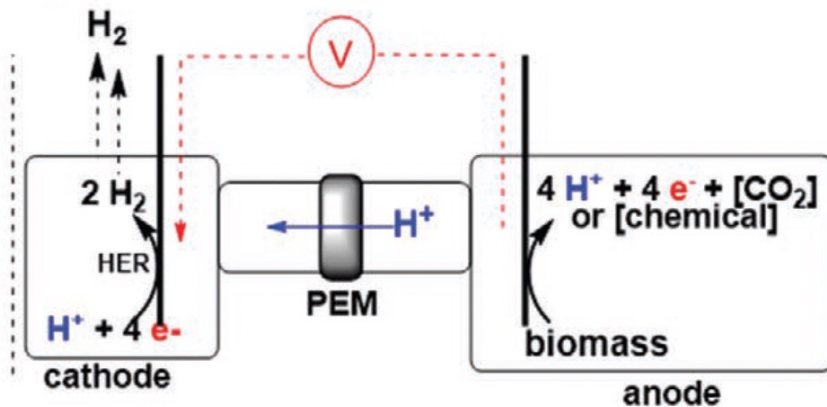


Green Hydrogen via Biomass Valorization



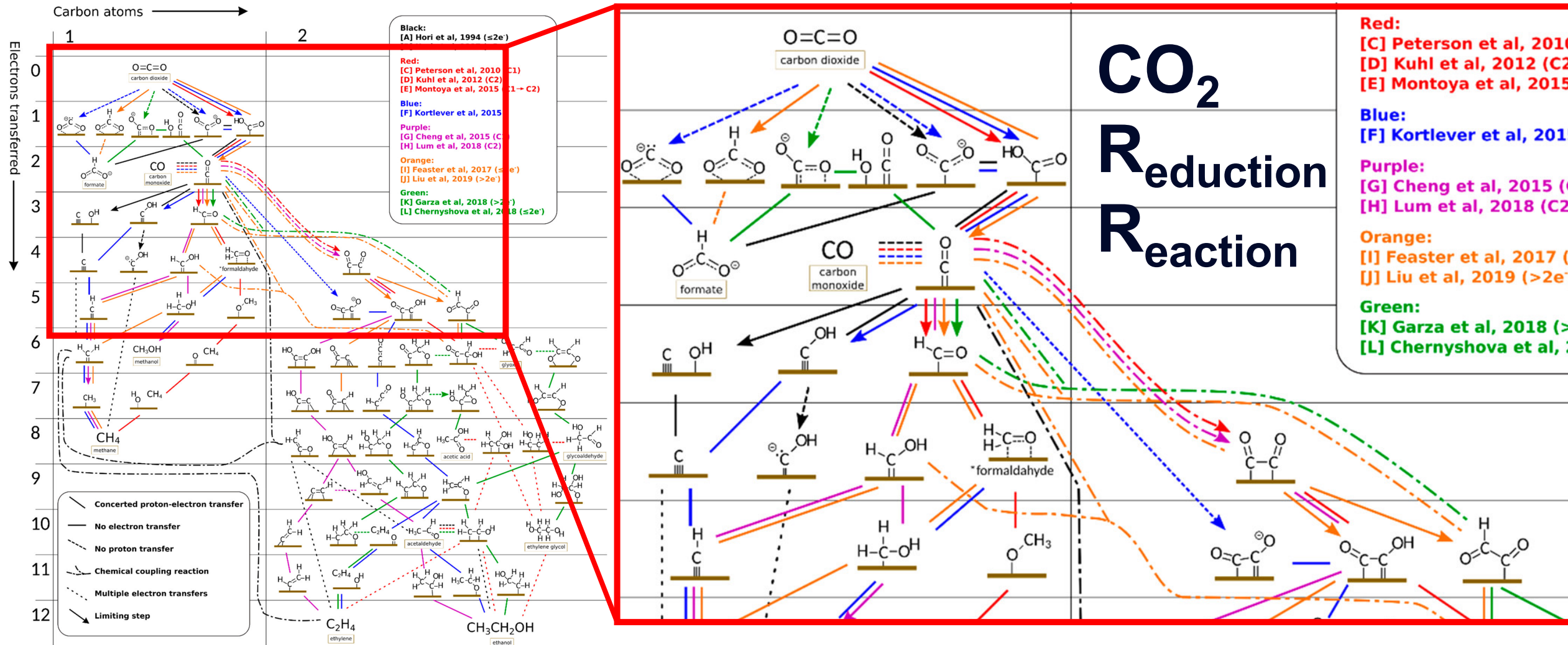
Advantage:

Much lower electricity demand
(almost 50%)

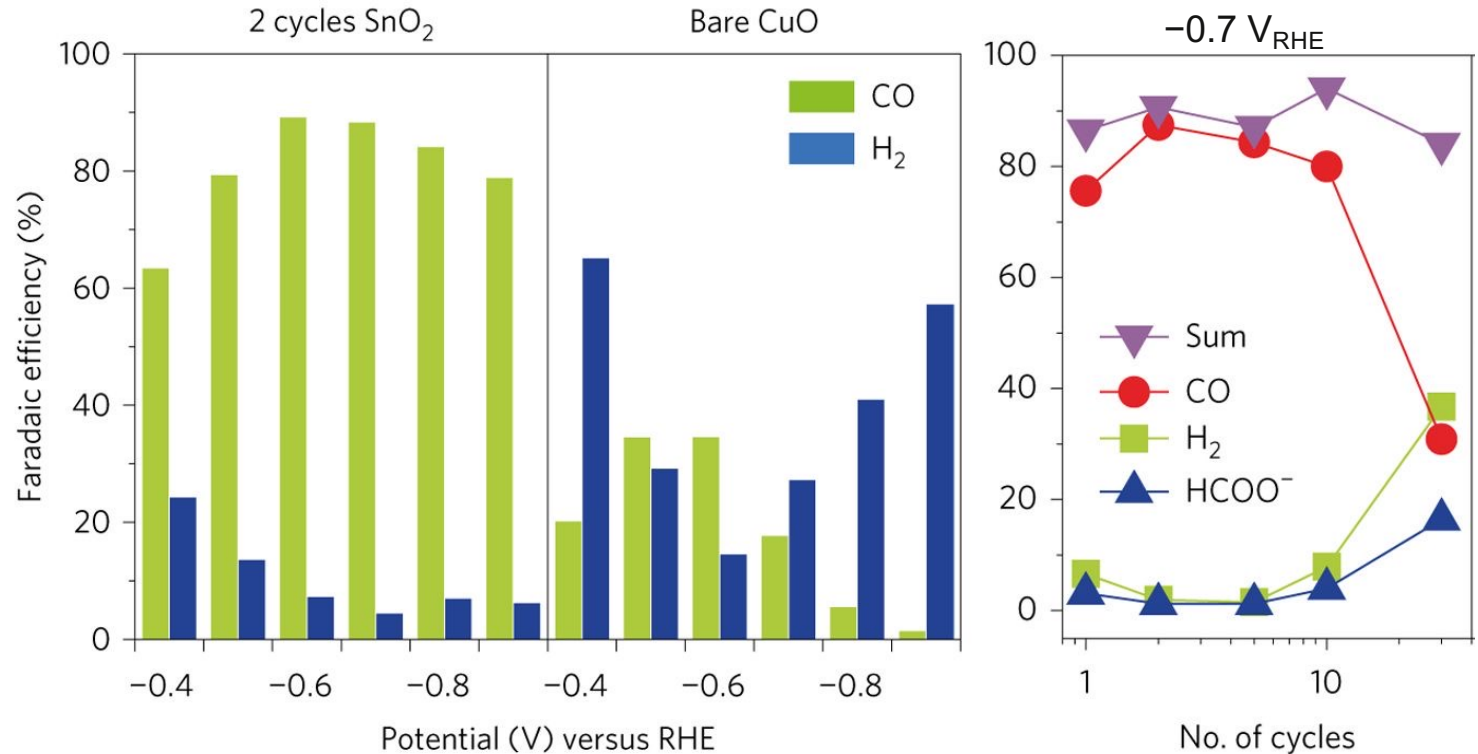
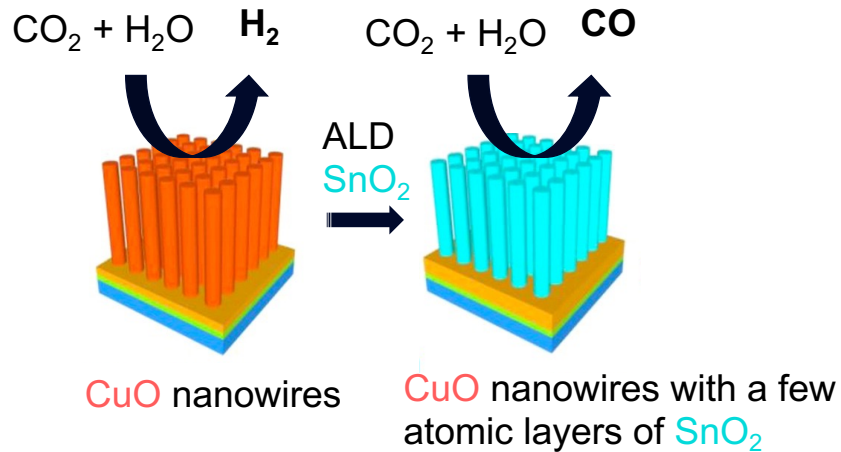


Challenge: Catalyst
selectivity towards partial
oxidation products

Green Hydrogen as a Co-product of CO₂RR



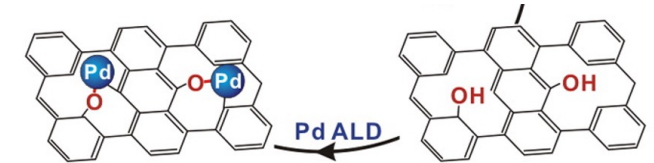
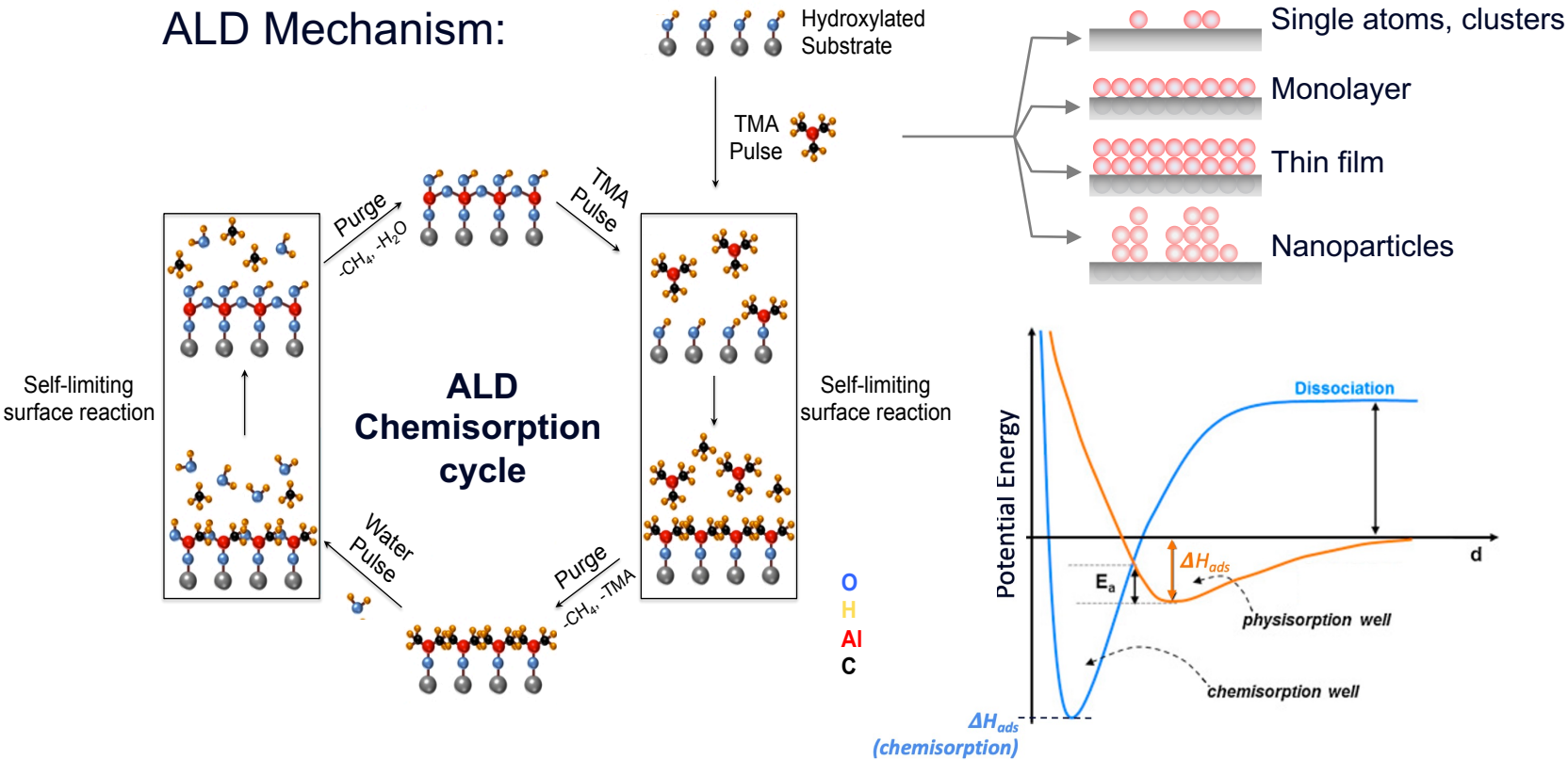
Importance of Single Atoms for Selectivity



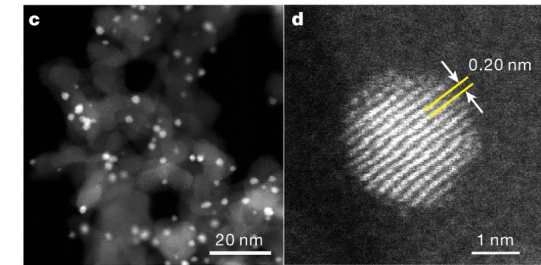
- Selectivity towards CO can switch from <20% to more than 70% with a single cycle of SnO₂ (less than a monolayer)
- Selectivity can be maintained between 80-85% for 2-10 cycles of SnO₂ (corresponding to <1.5 nm)
- When a dense pinhole-free film of SnO₂ begins to form (~30 cycles), selectivity towards CO drops sharply to 30% concomitant with increased production of formate and hydrogen

Nanoparticle and Single-atom catalysts with Atomic Layer Deposition

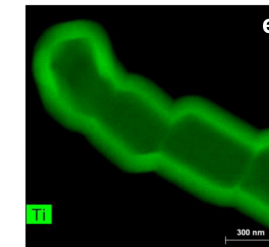
ALD Mechanism:



Yan, H. *et al.* *J Am Chem Soc* 137, 10484-10487 (2015).



Cao, L., Liu, W., Luo, Q. *et al.* *Nature* 565, 631–635 (2019)



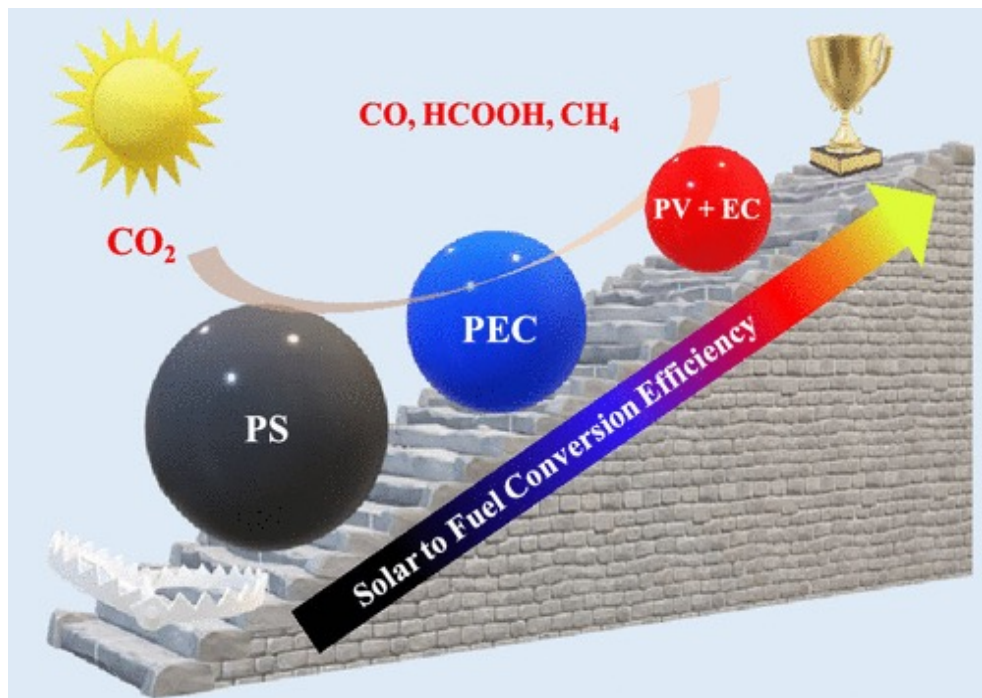
Luo, J., Steier L. *et al.* *Nano Lett* 16, 1848-1857 (2016).

Coating of porous high surface area substrates

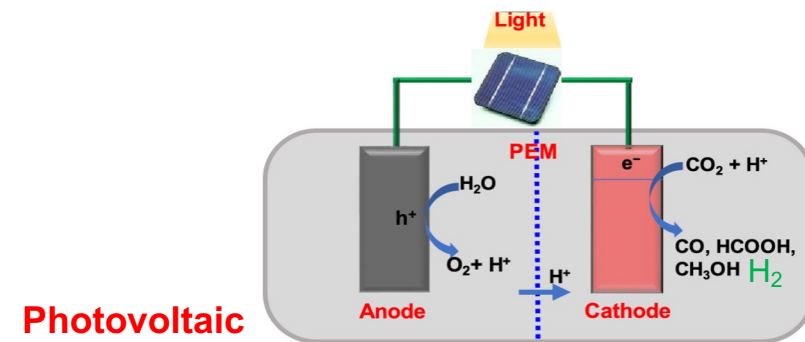
- Powerful tool to explore selectivity drivers and active site reactivity in catalysis
- Can help mitigating surface reconstruction of catalyst materials during operation

L. Steier, How to engineer surfaces to tailor selectivity?, in Stephens, I. E. L. *et al.* 2022 Roadmap on low temperature electrochemical CO₂ reduction. *Journal of Physics: Energy* 4, doi:10.1088/2515-7655/ac7823 (2022).

Routes to Green Hydrogen and Solar Fuels



$\eta_{\text{CO}_2\text{RR}}$	$\ll 1\%$	$\leq 10\%$	$\leq 19\%$
η_{HER}	$\leq 1\%$	$\leq 19\%$	$\leq 30\%$



Photovoltaic + electrochemical

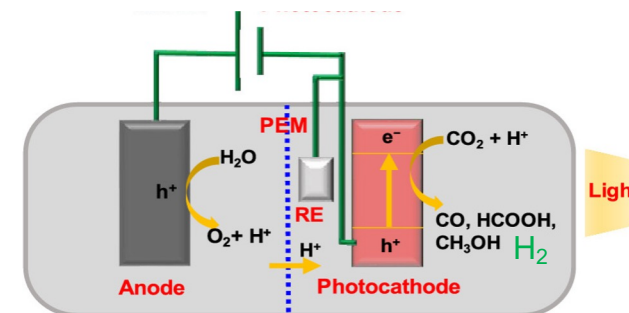
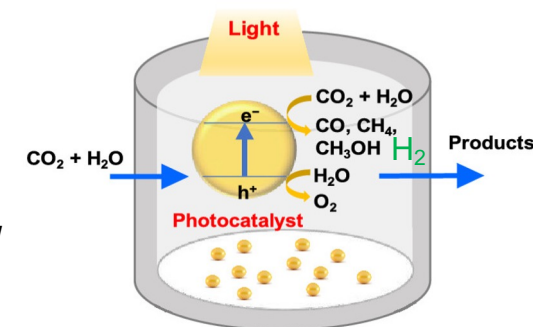


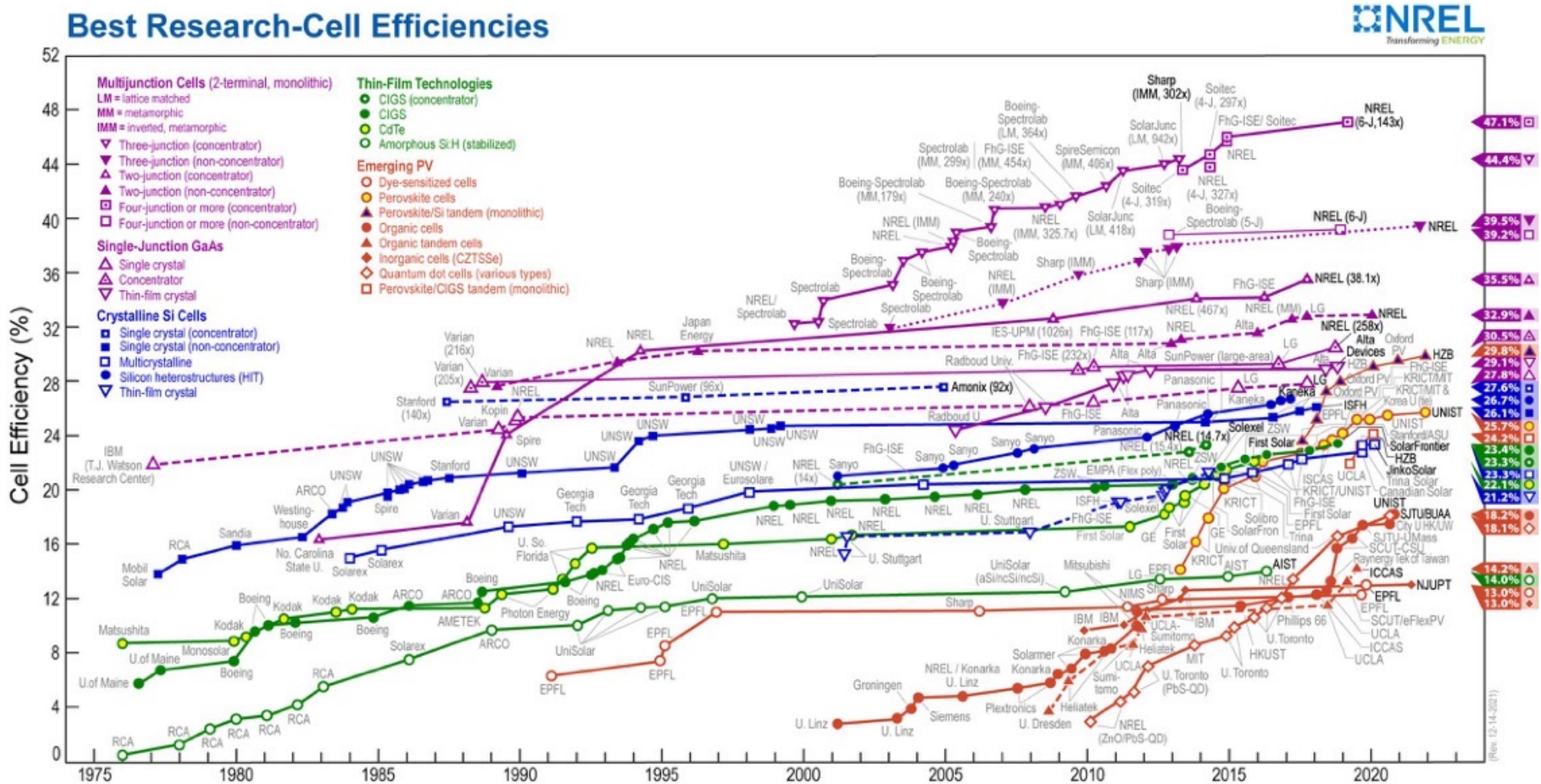
Photo-electrochemical



Photosynthetic/ photocatalytic

PV-Electrolysis

Overall efficiency = Efficiency of PV x Efficiency of Electrolyser



High carrier mobilities and long lifetimes in $\text{Cu}(\text{In},\text{Ga})\text{Se}_2$



Dr Y. Chang

solar cells
provided
by the group of:



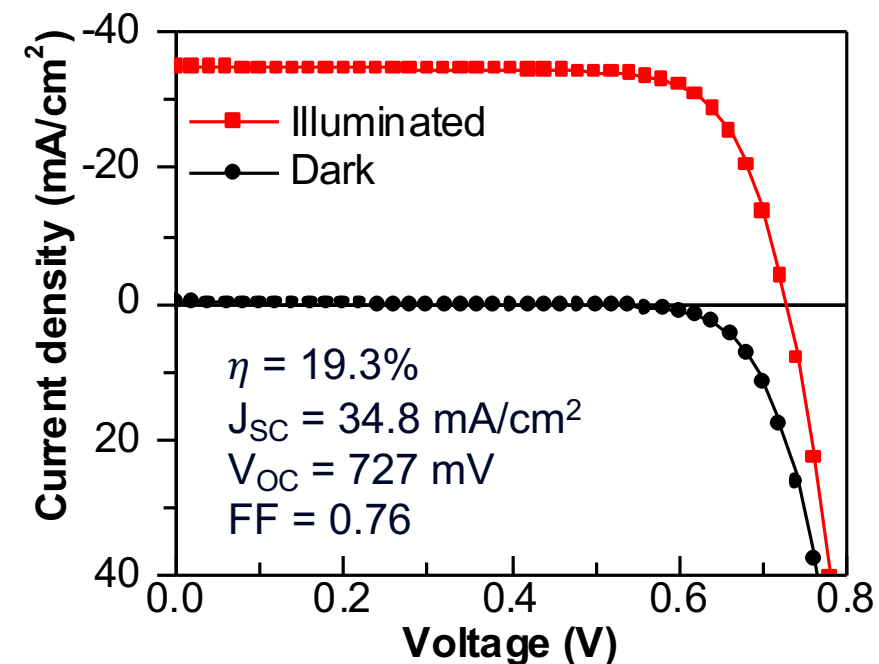
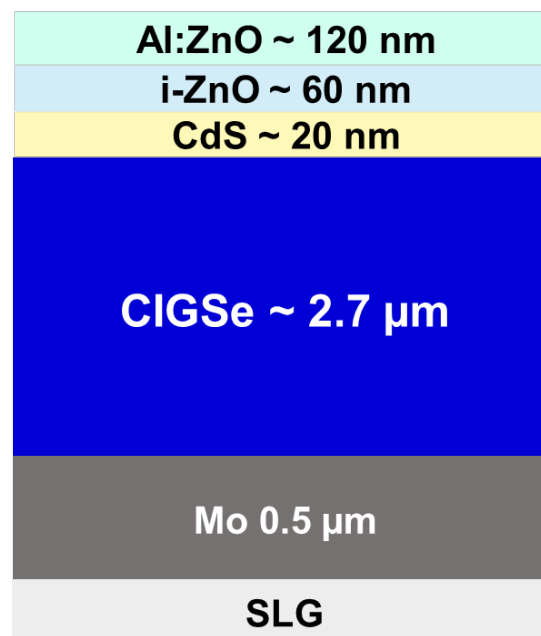
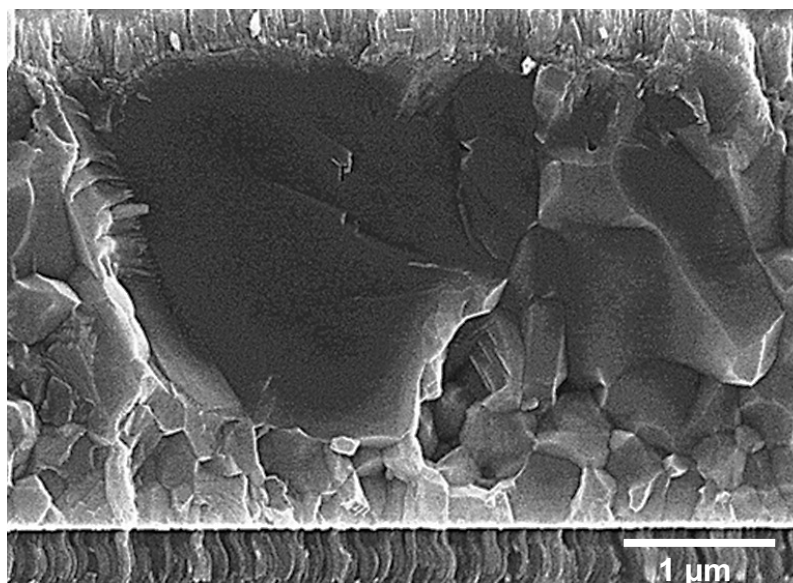
A. Tiwari



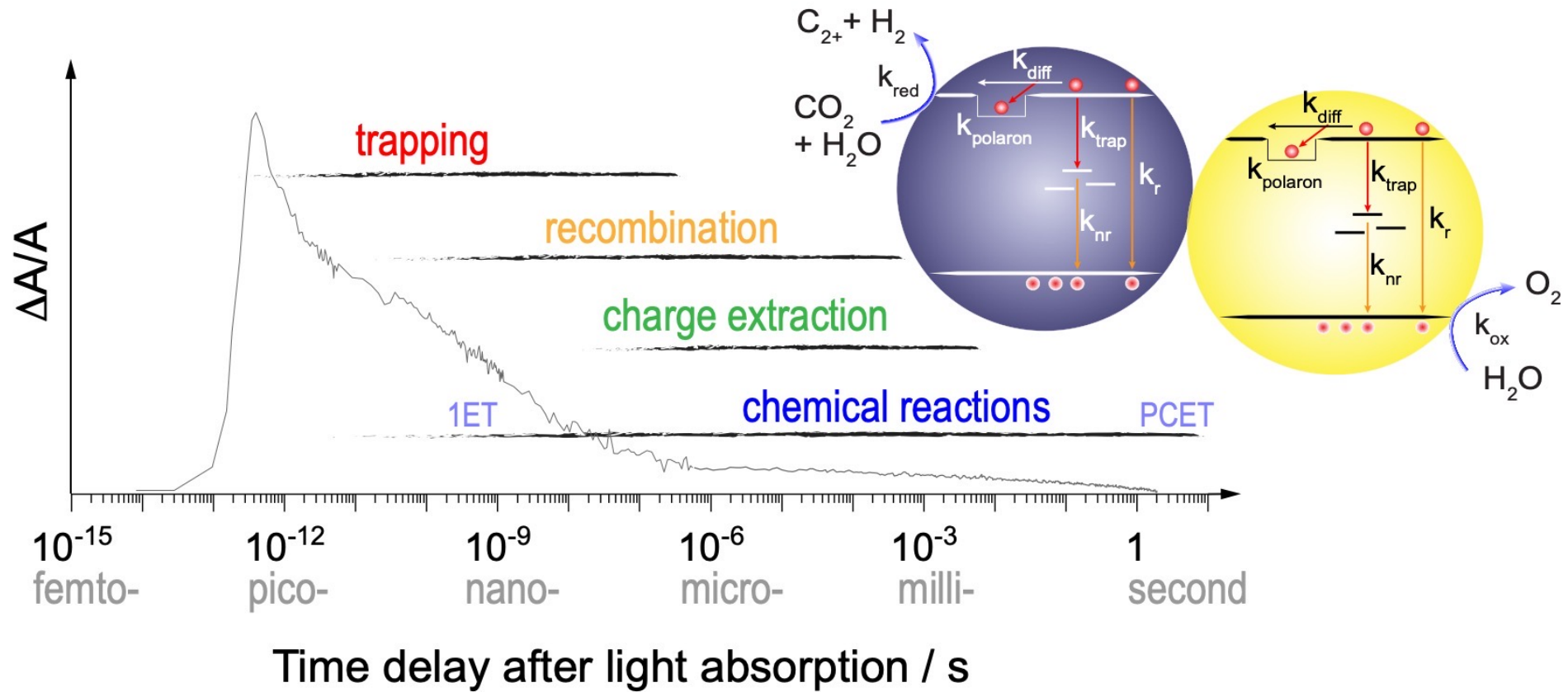
Empa

Materials Science and Technology

>19 ns carrier lifetimes in high-efficiency solar cells



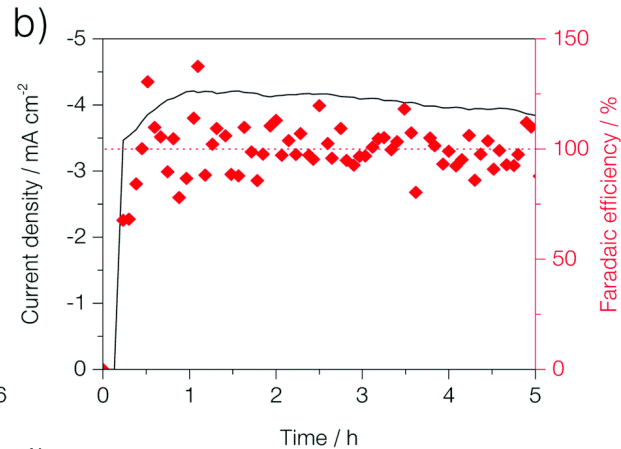
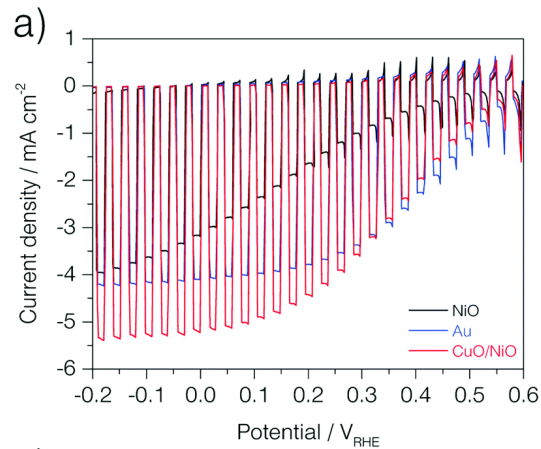
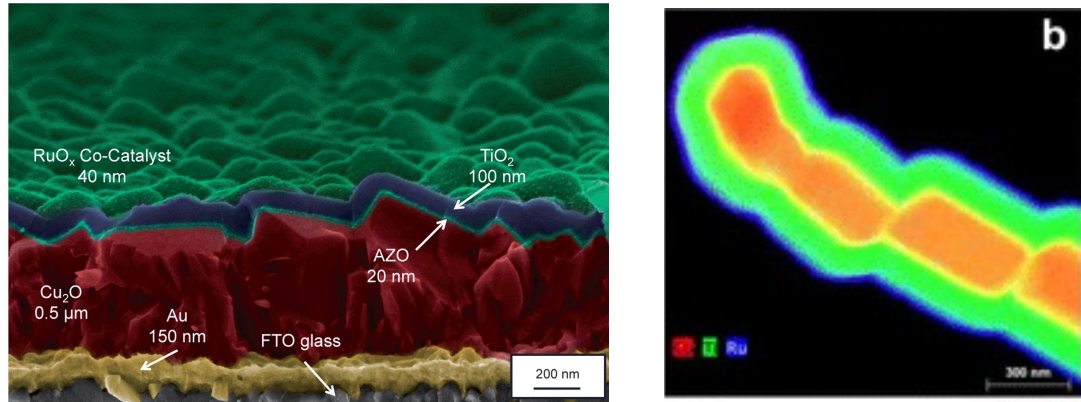
Kinetic Dilemma in Oxide Photocatalysts



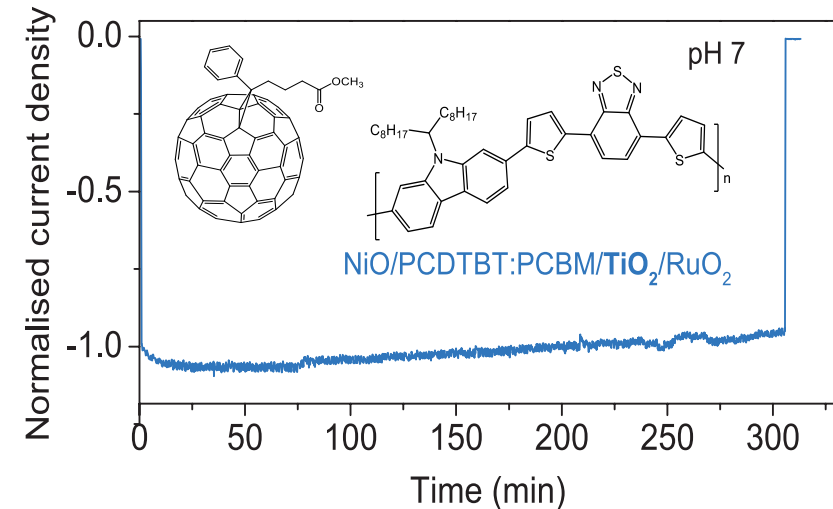
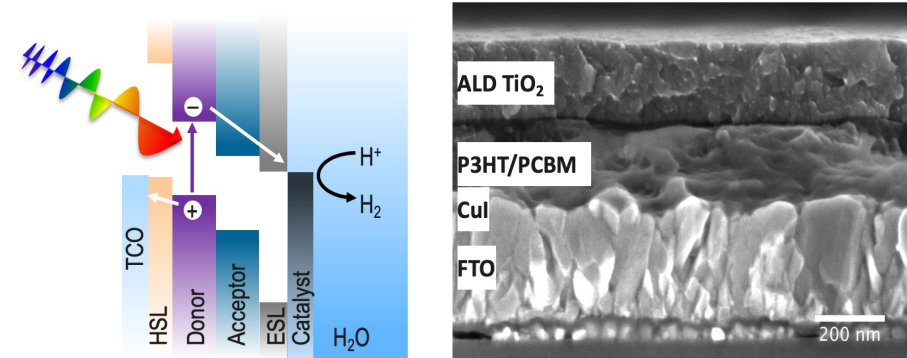
yield of photogenerated charge carriers on long timescales determines activity

Buried Junctions with TiO₂ overlayers

Cu₂O photocathodes



Polymer bulkheterojunction photocathodes



J. Azevedo, L. Steier *et al.* *Energy Environ. Sci.*, **7**, 4044-4052 (2014)
 J. Luo, L. Steier *et al.* *Nano Lett* **16**, 1848-1857 (2016)
 M. K. Son, L. Steier* *et al.* *Energy & Environmental Science* **10**, 912-918 (2017).

L. Steier*, *et al.* *Sustainable Energy & Fuels* **2017**, **1**, 1915-1920
 L. Francàs, E. Burns, L. Steier *et al.* *Chemical Commun.* **2018**, **54**, 5732-5735
 L. Steier*, S. Holliday, *J. Mater. Chem. A* **2018**, **6**, 21809-2182

The Photocatalyst Sheet Device

nature
materials

LETTERS

PUBLISHED ONLINE: 7 MARCH 2016 | DOI: 10.1038/NMAT4589

Scalable water splitting on particulate photocatalyst sheets with a solar-to-hydrogen energy conversion efficiency exceeding 1%

Qian Wang^{1,2}, Takashi Hisatomi^{1,2}, Qingxin Jia^{1,2}, Hiromasa Tokudome^{2,3}, Miao Zhong^{1,2}, Chizhong Wang¹, Zhenhua Pan¹, Tsuyoshi Takata⁴, Mamiko Nakabayashi⁵, Naoya Shibata⁵, Yanbo Li⁶, Ian D. Sharp⁶, Akihiko Kudo⁷, Taro Yamada^{1,2} and Kazunari Domen^{1,2*}

nature
energy

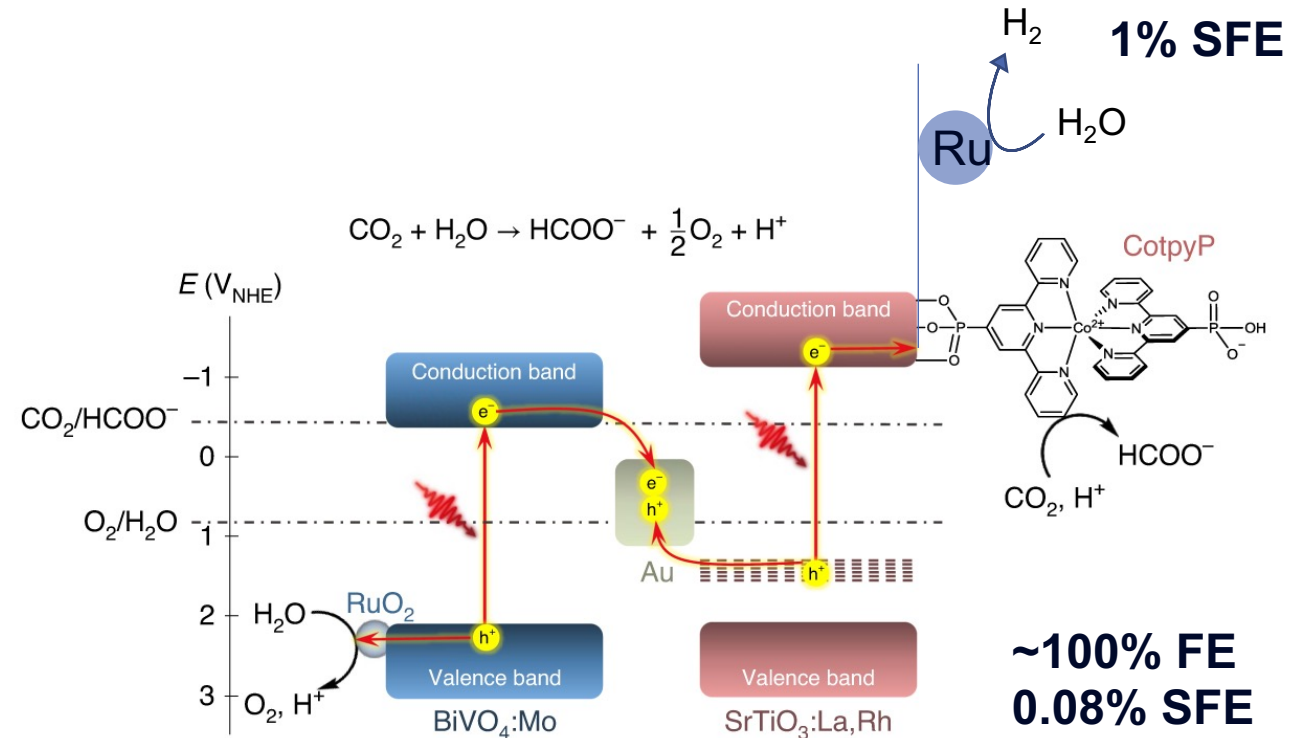
ARTICLES

<https://doi.org/10.1038/s41560-020-0678-6>

Check for updates

Molecularly engineered photocatalyst sheet for scalable solar formate production from carbon dioxide and water

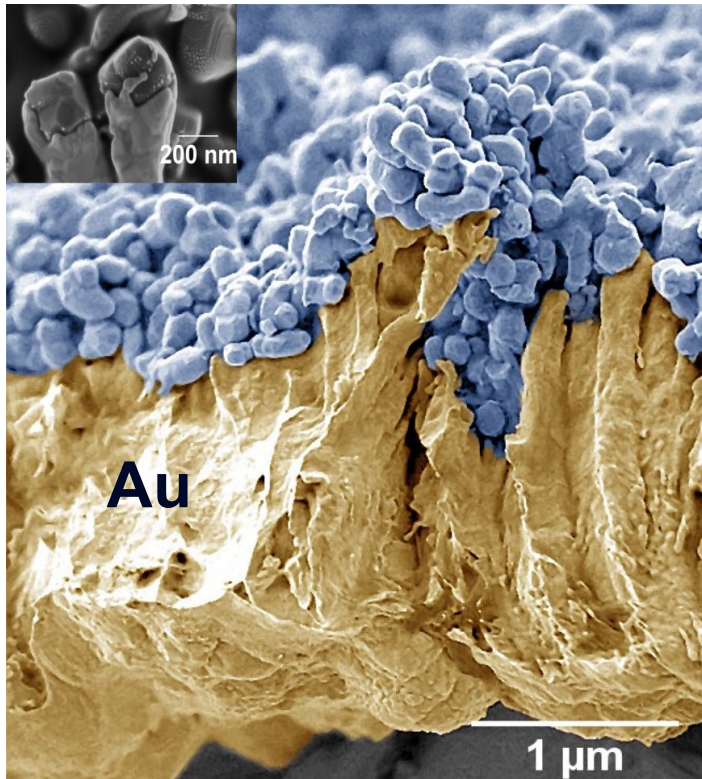
Qian Wang¹, Julien Warnan¹, Santiago Rodríguez-Jiménez¹, Jane J. Leung¹, Shafeer Kalathil¹, Virgil Andrei¹, Kazunari Domen^{2,3} and Erwin Reisner^{1✉}



Wang, Q. *et al.* *Nature Energy*, **5**, 703–710 (2020). doi:10.1038/s41560-020-0678-6.

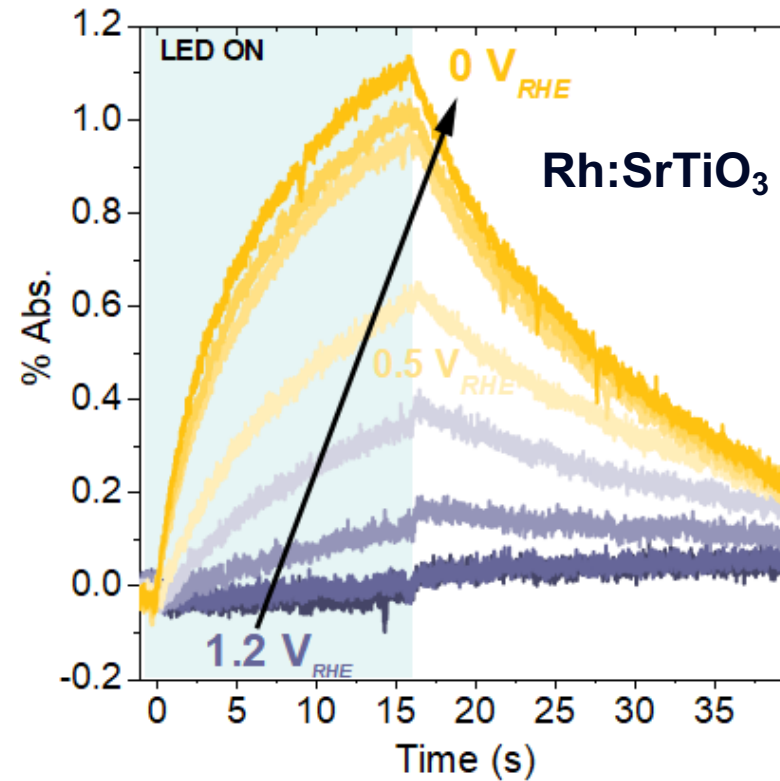
Wang, Q. *et al.* *Nature Materials* **15**, 611–+, (2016) doi:10.1038/nmat4589

Long electron lifetimes in La,Rh-doped SrTiO₃

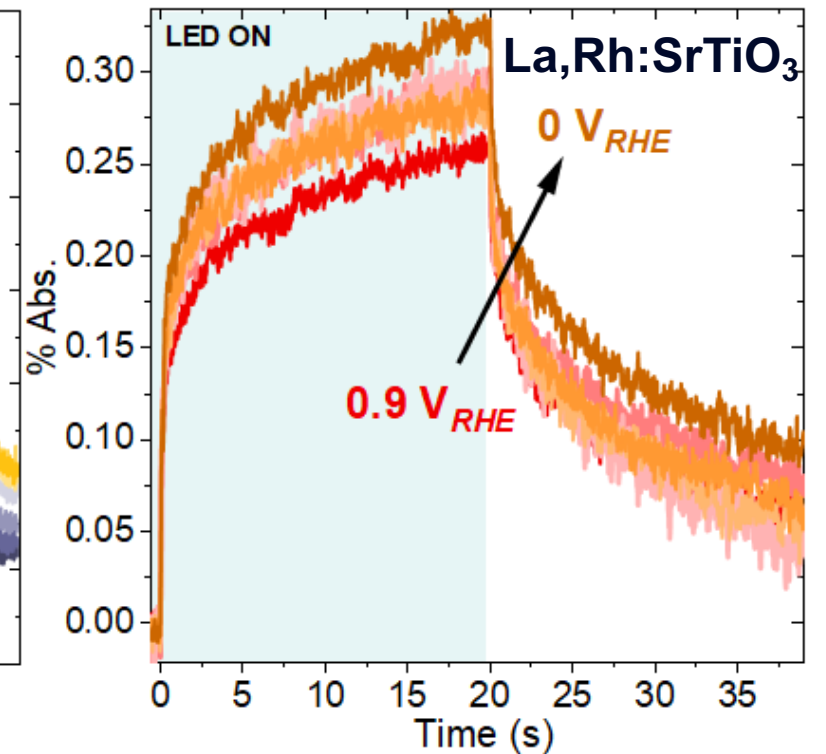


Photocatalyst sheet
half-electrode

Photoinduced absorption measurements
365 nm LED excitation, probe at 1100 nm (electrons)

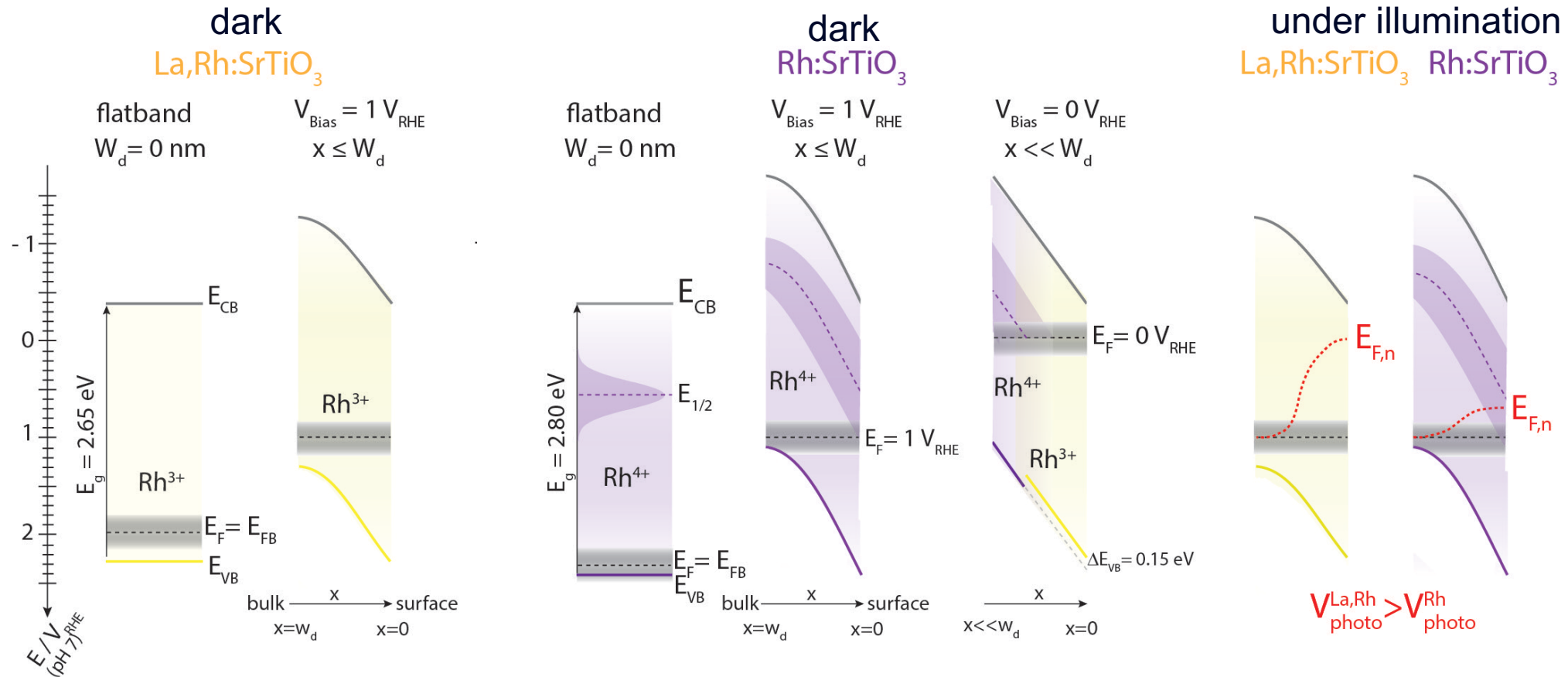


Rh⁴⁺
strong potential dependence



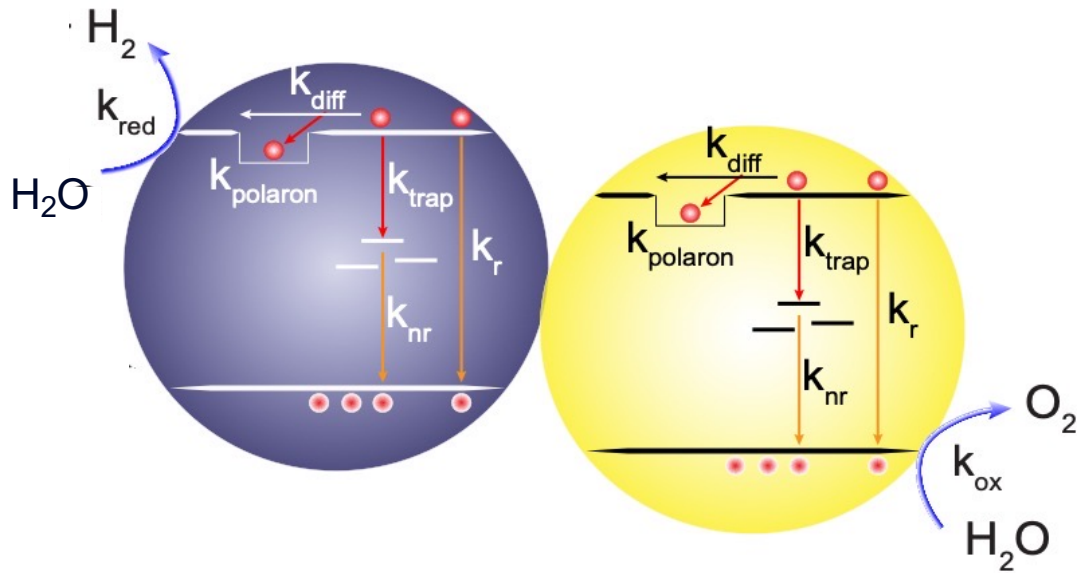
Rh³⁺
(almost) no potential dependence

Long electron lifetimes in La,Rh-doped SrTiO₃

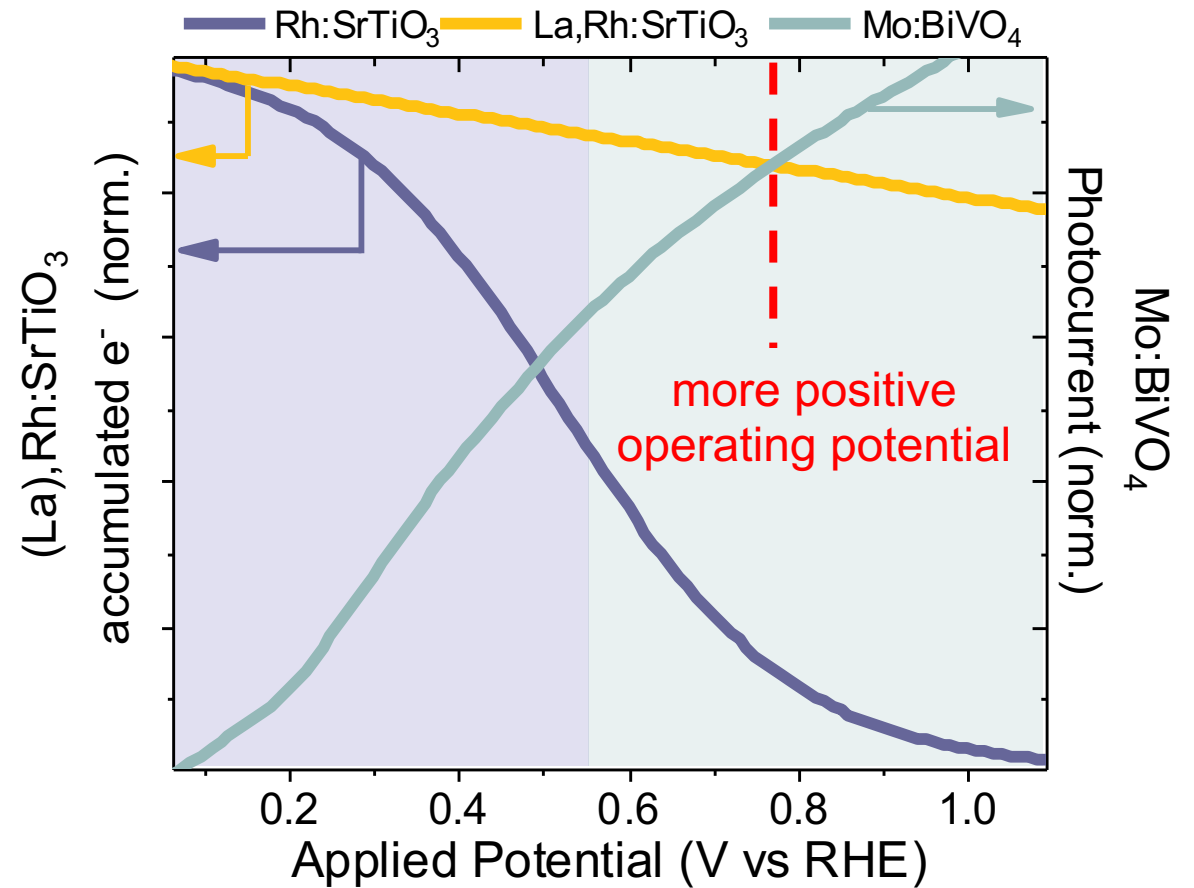


La co-doping reduces Rh^{4+} centres to Rh^{3+} - removing mid-bandgap Rh^{4+} recombination centres
 → enables accumulation of persistent electrons even under positive applied potentials
 → enables larger quasi-Fermi level splitting (photovoltage generation)

Operation of Mo:BiVO₄-La,Rh:SrTiO₃ Photocatalyst Sheet Device



Defect engineering allows the full sheet to operate at remarkably positive potentials



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